Received 21 June 2006

Accepted 22 June 2006

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Shan Liu,^a Jing-Ning Liu,^b Guang-Liang Song,^a Ya-Ming Wu^a and Hong-Jun Zhu^a*

^aDepartment of Applied Chemistry, College of Science, Nanjing University of Technology, Nanjing 210009, People's Republic of China, and ^bDepartment of Public Securital Science, Jiangsu Police Institute, Nanjing 210012, People's Republic of China

Correspondence e-mail: zhuhj@njut.edu.cn

Key indicators

Single-crystal X-ray study T = 315 K Mean σ (C–C) = 0.009 Å Disorder in solvent or counterion R factor = 0.086 wR factor = 0.229 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

3,7-Dichloro-3,7-diphenylbenzo[1,2-c;4,5-c']difuran-1,5(3*H*,7*H*)-dione cyclohexane 0.25-solvate

In the main molecule of the title compound, $C_{22}H_{12}Cl_2O_{4}$. 0.25 C_6H_{12} , the rings are planar. The asymmetric unit contains only one-half of a half-occupancy cyclohexane ring (site symmetry 111) and has a distorted chair conformation. Intramolecular $C-H\cdots Cl$ and $C-H\cdots O$ hydrogen bonds may be effective in the stabilization of the crystal structure.

Comment

Pseudo-2,5-dibenzoylterephthaloyl chloride is an intermediate used to synthesize the monomer 2,5-dibenzoyl-1,4-phenylenediamine, which can be utilized to synthesize organic semiconductors and conjugated polymers (Tonzola *et al.*, 2003). We report here the crystal structure of the title compound, (I).



In (I) (Fig. 1), the bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The rings *A* (C1–C6), *B* (O1/C7–C9/C14), *C* (C9–C14), *D* (O4/C11/C12/C15/C16) and *E* (C17–C22) are each planar; the dihedral angles between them are A/B = 73.40 (4)°, B/C = 1.77 (3)°, B/D = 5.30 (3)° and C/D = 3.54 (3)°. The asymmetric unit contains only one-half of the half-occupancy cyclohexane ring (site symmetry 111) and has a distorted chair conformation with the puckering parameters $Q_{\rm T} = 1.1050$ (3) Å, $\theta = 69.75$ (2)° and $\varphi = -48.11$ (3)° (Cremer & Pople, 1975).

As can be seen from the packing diagram (Fig. 2), the solvent molecules fill the cavities between the 2,5-dibenzoyl-terephthaloyl chloride (DBTC) molecules. The intramolecular $C-H\cdots Cl$ and $C-H\cdots O$ hydrogen bonds (Table 1) may influence the molecular conformation.

Experimental

© 2006 International Union of Crystallography All rights reserved The title compound was prepared from a mixture of 2,5-dibenzoylterephthalic acid (5.463 g, 14.6 mmol) (Liu *et al.*, 2006), thionyl

organic papers

chloride (21.0 ml) and *N*,*N*-dimethylformamide (0.10 ml) as catalyst, heated with stirring at 349–351 K for 2 h. From the resulting clear solution, the excess thionyl chloride was removed by distillation. Benzene (10 ml) was added and the mixture was distilled to remove the last traces of thionyl chloride. DBTC was then obtained as fine white leaflets (yield 5.922 g, 98.6%). The crystals were obtained by dissolving DBTC (1.0 g) in cyclohexane (100 ml) and evaporation of the solvent at room temperature over a period of about 45 d.

Z = 8

 $D_x = 1.321 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.33 \text{ mm}^{-1}$ T = 315 (2) K

Block, colorless $0.30 \times 0.20 \times 0.20$ mm

 $\theta_{\rm max} = 26.0^{\circ}$

4254 independent reflections

3 standard reflections

frequency: 120 min

intensity decay: none

1954 reflections with $I > 2\sigma(I)$

Crystal data

C22H12Cl2O4·0.25C6H12
$M_r = 432.26$
Orthorhombic, Pbca
a = 10.817 (2) Å
b = 13.679 (3) Å
c = 29.370 (6) Å
$V = 4345.8 (15) \text{ Å}^3$

Data collection

Enraf–Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{min} = 0.909, T_{max} = 0.938$ 4254 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.07P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.086$	+ 6P]
$wR(F^2) = 0.229$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.11	$(\Delta/\sigma)_{\rm max} = 0.007$
4254 reflections	$\Delta \rho_{\rm max} = 0.78 \ {\rm e} \ {\rm \AA}^{-3}$
280 parameters	$\Delta \rho_{\rm min} = -0.29 \text{ e} \text{ \AA}^{-3}$
H-atom parameters constrained	

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$	
C6-H6A···Cl1	0.93	2.70	3.076 (6)	105	
C18-H18A···O4	0.93	2.45	2.775 (9)	101	
$C22 - H22A \cdot \cdot \cdot Cl2$	0.93	2.70	3.069 (7)	104	

H atoms were positioned geometrically, with C-H = 0.93 and 0.97 Å for aromatic and methylene H atoms, respectively, and constrained to ride on their parent atoms, with $U_{iso}(H) = 1.2U_{eq}(C)$. The cyclohexane molecule was refined successfully with half-occupancy. There is probably unresolved disorder, as indicated by the displacement ellipsoids, which makes the ring appear more nearly planar than it really is.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1985); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL*.

The authors thank the Center of Test and Analysis, Nanjing University, for support.



Figure 1

The molecular structure of (I), with the atom numbering scheme. Displacement ellipsoids are drawn at the 30% probability level [symmetry code: (A) $\frac{1}{2} - x$, -y, $\frac{1}{2} + z$]. Hydrogen bonds are shown as dashed lines.



Figure 2

A partial packing diagram of (I). The intramolecular hydrogen bonds are shown as dashed lines.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Bruker (2000). XSCANS and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Enraf–Nonius (1985). *CAD-4 Software*. Version 5.0. Enraf–Nonius, Delft, The Netherlands.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.
- Liu, S., Ji, J.-X., Wang, D.-D. & Zhu, H.-J. (2006). Acta Cryst. E62, o1387-01389.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351– 359.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Tonzola, C. J., Alam, M. M., Kaminsky, W. & Jenekhe, S. A. (2003). J. Am. Chem. Soc. 125, 13548–13558.